

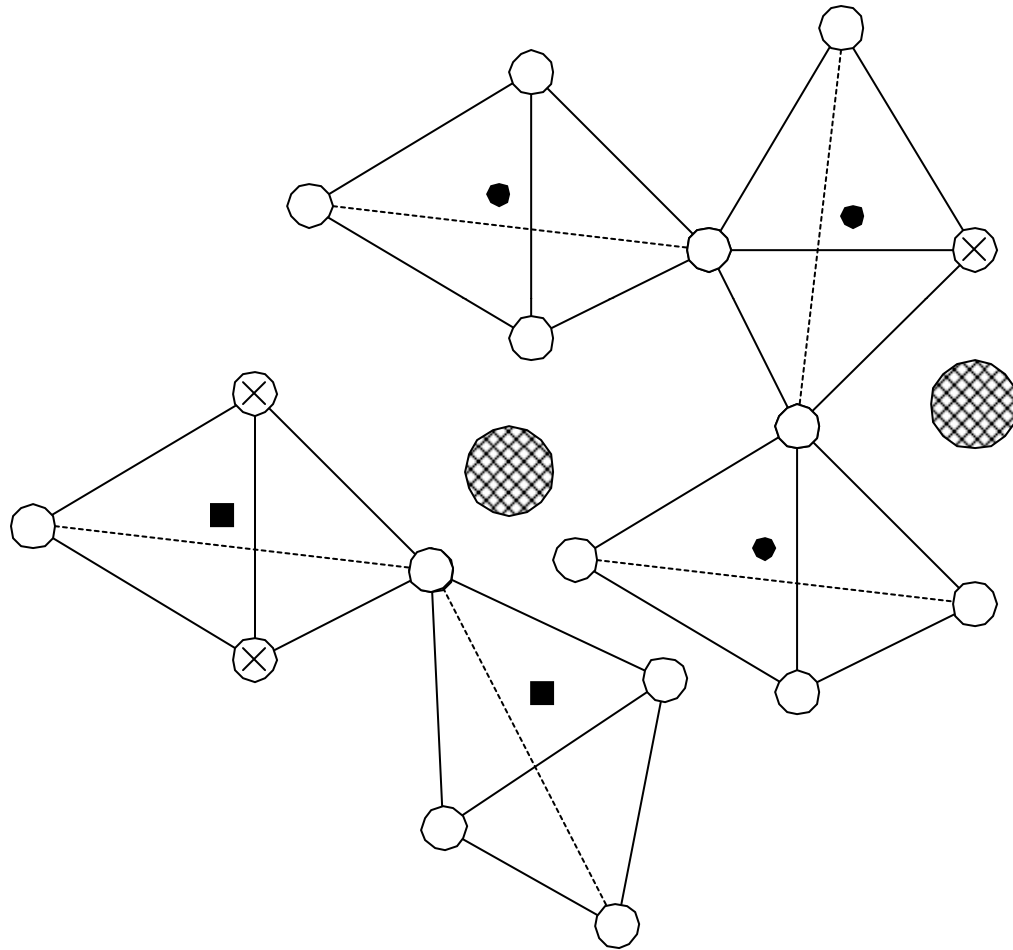
# Silicon dioxide






- What is  $\text{SiO}_2$ ?
- What is  $\text{SiO}_2$  used for?
- Advantages and Disadvantages of  $\text{SiO}_2$
- How is it grown?
  - Dry
  - Wet
- Numerical Examples

# What is SiO<sub>2</sub>?

- Two forms
  - Single crystal (quartz)
  - Amorphous
- We are interested in Amorphous SiO<sub>2</sub>
  - Random three dimensional network of SiO<sub>2</sub> constructed from polyhedra of oxygen ions.
  - This material is more porous than Quartz (density of 2.15-2.25g/cm<sup>3</sup> compared to 2.6525g/cm<sup>3</sup> )

# What is SiO<sub>2</sub>?



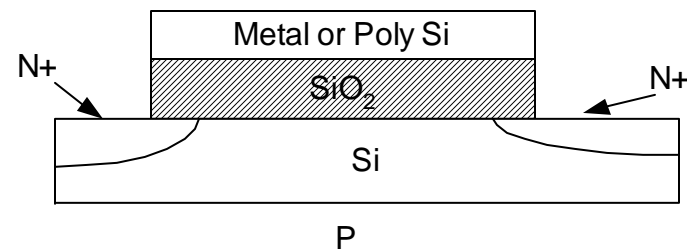
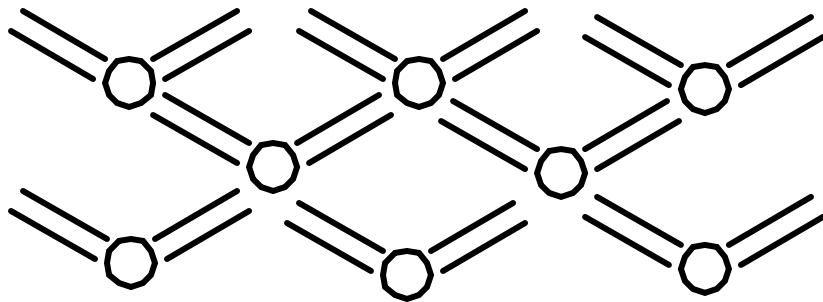
-  Network modifier
-  Nonbridging oxygen
-  Silicon
-  Network former
-  Bridging oxygen

The O-Si-O Bond angle is 109°

Tetrahedral distance between Si and O ions is 1.6Å

# What is $\text{SiO}_2$ used for?

- MOS Metal Oxide Semiconductor
- Device passivation
  - Combines with dangling bonds to reduce surface states

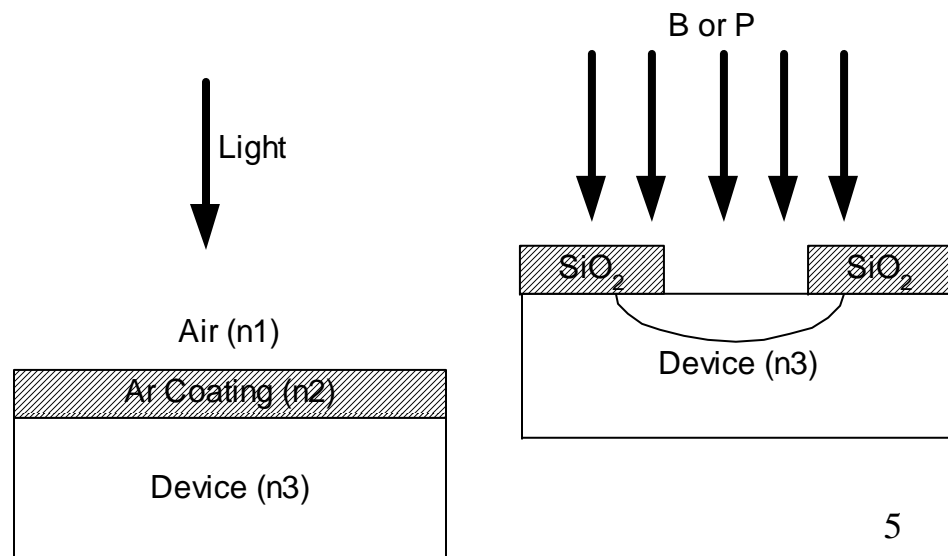


# What is SiO<sub>2</sub> used for?

- Diffusion Masks
  - Block the diffusion of B and P for example
- Antireflective coating for Photodevices

$$n_2 = \sqrt{n_1 n_3},$$

$$\text{thickness} = \frac{\lambda}{4n_2}$$



# Advantages and Disadvantages of $\text{SiO}_2$

- CMOS digital logic gates use little power when not switching logic state, thus high levels of integration are possible because the standby power consumption is low.
- $\text{SiO}_2$  is a native film that is quite easy to grow. All that is required is heat and oxygen or steam.

# Advantages and Disadvantages of $\text{SiO}_2$

- $\text{SiO}_2$  consumes Si while growing. 44% of the  $\text{SiO}_2$  layer comes from the original Si.
  - This leads to a non-planer structure after each oxidation step.
- Due to the large increase in volume there is  $2\text{-}4 \times 10^9$  dyn  $\text{cm}^{-1}$  of compressive strain.
  - This causes dislocations.
- Oxidation-Induced Stacking Faults (these can be removed by a high temp treatment.

# Advantages and Disadvantages of $\text{SiO}_2$

- The large dielectric constant leads to larger capacitance values for a given thickness (compared to silicon nitride).

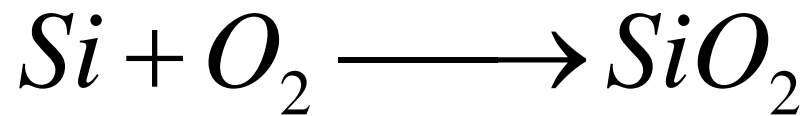
# How is it grown?

- The oxidizing species must diffuse through the SiO<sub>2</sub> layer that has already grown. This leads to a linear regime of growth and a parabolic regime of growth. Given by the equation:

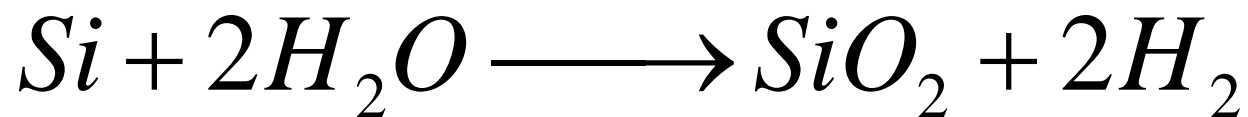
$$X^2 + A(\mu m)X = B(\mu m^2 / hr)t(hr)$$

# How is it grown?

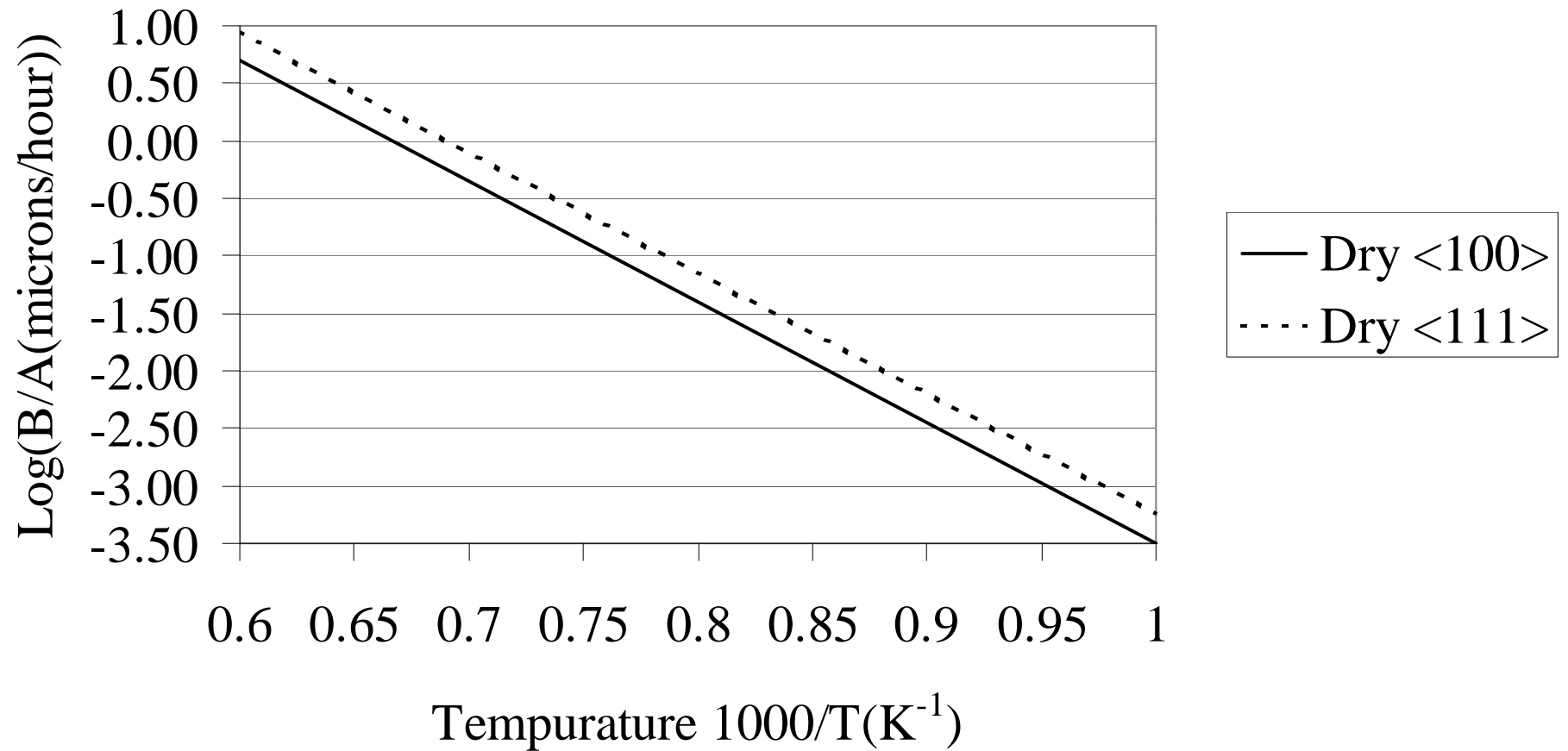
- Dry oxidation: Flow dry O<sub>2</sub> over sample at elevated temperatures.



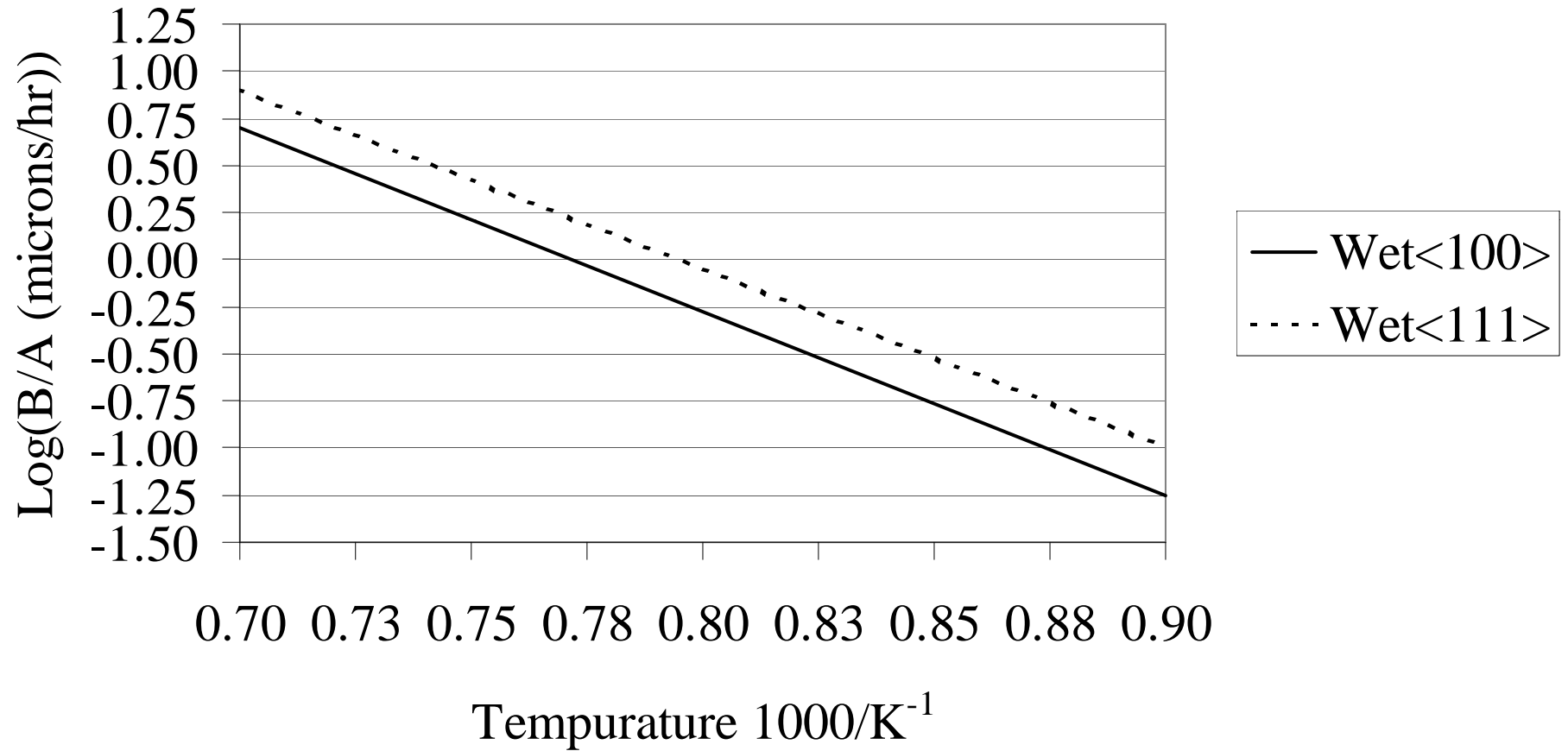
- Wet oxidation: Bubble N<sub>2</sub> through a water bubbler @95C° over sample at elevated temperatures.



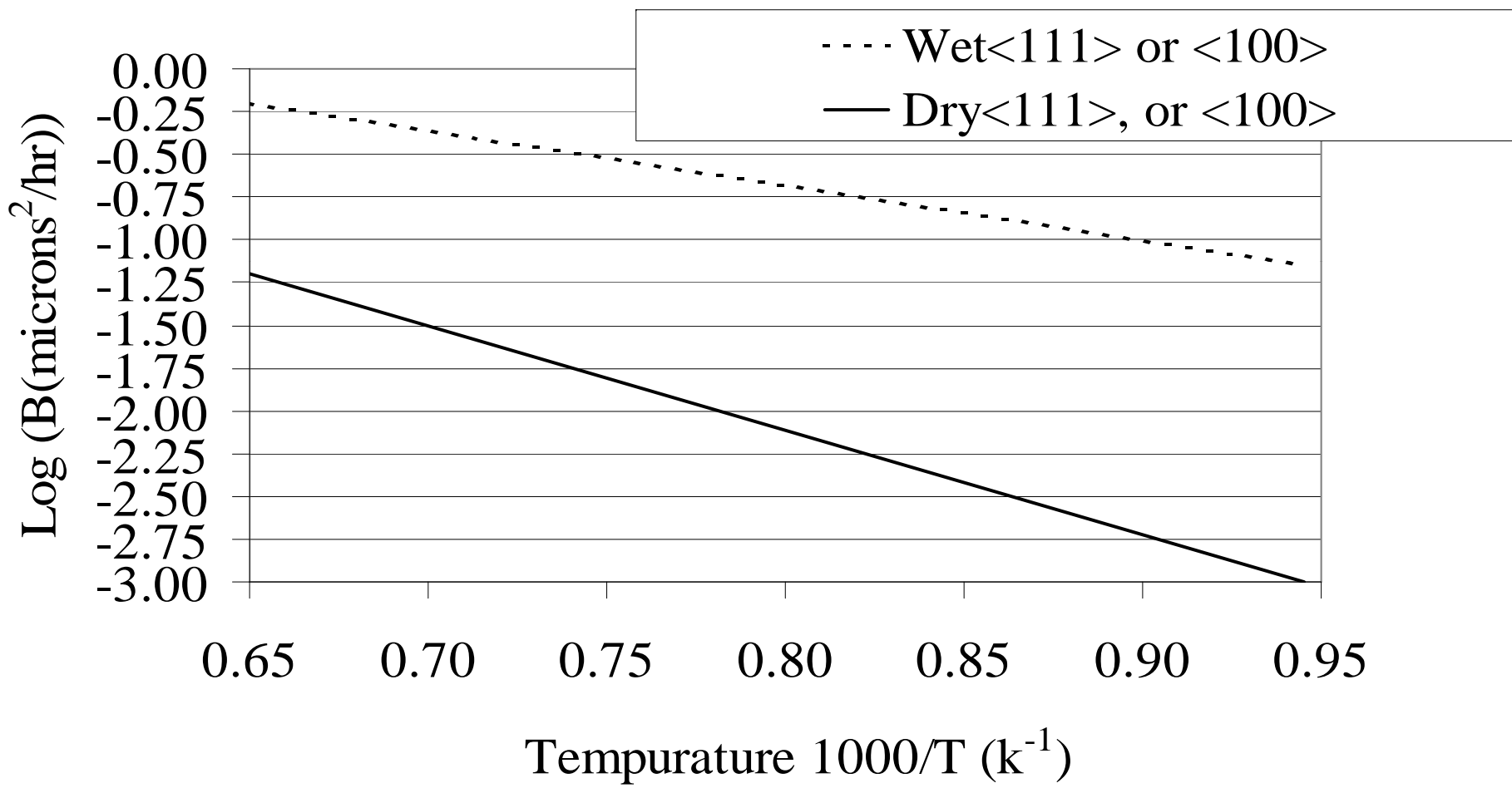
## Linear Rate Constant versus Temperature



# Wet Oxidation



Parabolic Rate Constant versus Temperature



# Example 1

- Need to grow an Anti Reflective Coating for silicon based solar cells.
- Commercial cells are “blue” corresponding to a SiO<sub>2</sub> thickness  $(X_o)=0.1\mu\text{m}$ .
- The after orientation is  $\langle 111 \rangle$  due to the fact  $\langle 111 \rangle$  Silicon has a longer lifetime, thus a longer diffusion length and thus more efficiency cells than  $\langle 100 \rangle$

# Choose wet/dry process and temperature

- We will choose 1150°C and wet because this should minimize our process time.

	Wet Oxidation		
<100> Silicon	Do		
Linear (B/A)	9.70E+07	um/hr	2.05eV
Parabolic (B)	386	um <sup>2</sup> /hr	0.78eV
<111> Silicon			
Linear (B/A)	1.63E+08	um/hr	2.05eV
Parabolic (B)	3.86E+02	um <sup>2</sup> /hr	0.78eV

	Dry Oxidation		
<100> Silicon			
Linear (B/A)	3.71E+06	um/hr	2eV
Parabolic (B)	772	um <sup>2</sup> /hr	1.23eV
<111> Silicon			
Linear (B/A)	6.23E+06	um/hr	2eV
Parabolic (B)	7.72E+02	um <sup>2</sup> /hr	1.23eV

# First Pass

$$k := 8.62 \cdot 10^{-5} \frac{\text{V}}{\text{K}} \quad \text{wet } \langle 111 \rangle \text{ Si}$$

$$\frac{B}{A} = D_o \cdot \exp\left(\frac{-E_A}{kT}\right)$$

$$D_o := 1.63 \cdot 10^8 \frac{\mu\text{m}}{\text{hr}} \quad E_A := 2.05\text{V}$$

$$T := (1150 + 273)\text{K} \quad \frac{-E_A}{k \cdot T} = -16.713$$

$$\text{Linear} = \frac{B}{A}$$

$$\text{Linear} := D_o \cdot \exp\left(\frac{-E_A}{k \cdot T}\right)$$

$$\text{Linear} = 8.996 \frac{\mu\text{m}}{\text{hr}}$$

$$B = D_o \cdot \exp\left(\frac{-E_A}{kT}\right)$$

$$D_o := 386 \frac{\mu\text{m}^2}{\text{hr}}$$

$$E_A := 0.78\text{V}$$

$$T := (1150 + 273)\text{K}$$

$$\frac{-E_A}{k \cdot T} = -6.359$$

$$B := D_o \cdot \exp\left(\frac{-E_A}{k \cdot T}\right)$$

$$B = 0.668 \frac{\mu\text{m}^2}{\text{hr}}$$

$$X_o := 0.1 \mu\text{m}$$

$$\text{Time} := \frac{X_o^2}{B} + \frac{X_o}{\text{Linear}} \quad \text{Time} = 0.026 \text{ hr}$$

Time too short for our furnace! 16

# Second Pass (900°C)

$$k := 8.62 \cdot 10^{-5} \frac{\text{V}}{\text{K}} \quad \text{wet } \langle 111 \rangle \text{ Si}$$

$$\frac{B}{A} = D_o \cdot \exp\left(\frac{-E_A}{kT}\right)$$

$$D_o := 1.63 \cdot 10^8 \frac{\mu\text{m}}{\text{hr}} \quad E_A := 2.05\text{V}$$

$$T := (900 + 273)\text{K} \quad \frac{-E_A}{k \cdot T} = -20.274$$

$$\text{Linear} = \frac{B}{A}$$

$$\text{Linear} := D_o \cdot \exp\left(\frac{-E_A}{k \cdot T}\right)$$

$$\text{Linear} = 0.255 \frac{\mu\text{m}}{\text{hr}}$$

$$B = D_o \cdot \exp\left(\frac{-E_A}{kT}\right)$$

$$D_o := 386 \frac{\mu\text{m}^2}{\text{hr}} \quad E_A := 0.78\text{V}$$

$$T := (900 + 273)\text{K} \quad \frac{-E_A}{k \cdot T} = -7.714$$

$$B := D_o \cdot \exp\left(\frac{-E_A}{k \cdot T}\right) \quad B = 0.172 \frac{\mu\text{m}^2}{\text{hr}}$$

$$X_o := 0.1 \mu\text{m}$$

$$\text{Time} := \frac{X_o^2}{B} + \frac{X_o}{\text{Linear}} \quad \text{Time} = 0.45 \text{ hr}$$

Time just right for our furnace!

$$\text{Time} = 26.98 \text{ min}$$

# Wet Oxide

				Wet Oxidation			
<100> Silicon				Do			
Linear (B/A)				9.70E+07 um/hr		2.05 eV	
Parabolic (B)				386 um <sup>2</sup> /hr		0.78 eV	
<111> Silicon							
Linear (B/A)				1.63E+08 um/hr		2.05 eV	
Parabolic (B)				3.86E+02 um <sup>2</sup> /hr		0.78 eV	
		Temp oC	Temp (K)	Do	Ea/KT	Do*exp(-Ea/kT)	
Example 1	B/A	1150	1423	1.63E+08	16.71251058	8.995731445	
.1um, 1150oC	B	1150	1423	3.86E+02	6.358906464	0.668265924	
Wet							
<111> Si							
		B/A	B	Thickness (um)		Time=Xo <sup>2</sup> /B+Xo/(B/A) (hr)	
		8.995731	0.668265924	0.1		0.026080486	
						Time(min)=	
						1.564829	

				Wet Oxidation			
<100> Silicon				Do			
Linear (B/A)				9.70E+07 um/hr		2.05 eV	
Parabolic (B)				386 um <sup>2</sup> /hr		0.78 eV	
<111> Silicon							
Linear (B/A)				1.63E+08 um/hr		2.05 eV	
Parabolic (B)				3.86E+02 um <sup>2</sup> /hr		0.78 eV	
		Temp oC	Temp (K)	Do	Ea/KT	Do*exp(-Ea/kT)	
Example 2	B/A	900	1173	1.63E+08	20.27442673	0.255338294	
.1um, 1150oC	B	900	1173	3.86E+02	7.714172121	0.1723316	
Wet							
<111> Si							
		B/A	B	Thickness (um)		Time=Xo <sup>2</sup> /B+Xo/(B/A) (hr)	
		0.255338	0.1723316	0.1		0.449664963	
						Time(min)=	
						26.9799	

# Dry Oxide Examples

<100> Silicon						
Linear (B/A)				3.71E+06	um/hr	2eV
Parabolic (B)				772	um <sup>2</sup> /hr	1.23eV
<111> Silicon						
Linear (B/A)				6.23E+06	um/hr	2eV
Parabolic (B)				7.72E+02	um <sup>2</sup> /hr	1.23eV
		Temp oC	Temp (K)	Do	Ea/KT	Do*exp(-Ea/kT)
Example 1	B/A	1150	1423	6.23E+06	16.30489	0.516850571
.1um, 1150oC	B	1150	1423	7.72E+02	10.02751	0.034097821
Dry						
<111> Si						
			B/A	B	Thickness (um)	Time=Xo <sup>2</sup> /B+Xo/(B/A) (hr)
			0.516851	0.034097821	0.1	0.486753392
					Time(min)=	29.2052

<100> Silicon						
Linear (B/A)				3.71E+06	um/hr	2eV
Parabolic (B)				772	um <sup>2</sup> /hr	1.23eV
<111> Silicon						
Linear (B/A)				6.23E+06	um/hr	2eV
Parabolic (B)				7.72E+02	um <sup>2</sup> /hr	1.23eV
		Temp oC	Temp (K)	Do	Ea/KT	Do*exp(-Ea/kT)
Example 2	B/A	900	1173	6.23E+06	19.77993	0.016001999
.1um, 1150oC	B	900	1173	7.72E+02	12.16466	0.004023225
Dry						
<111> Si						
			B/A	B	Thickness (um)	Time=Xo <sup>2</sup> /B+Xo/(B/A) (hr)
			0.016002	0.004023225	0.1	8.734787461
					Time(min)=	524.0872